

SAFETY DATA SHEET

1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER

1.1 Product identifier

Product name RED DIRT REMOVER

Synonyms

1.2 Uses and uses advised against Uses ACIDIC DETERGENT

1.3 Details of the supplier of the product

Supplier name	LIBERATO BULK CHEMICAL & REPACK SPECIALISTS PTY. LTD.
Address	1 Kalinga Way, Landsdale, WA, 6065, AUSTRALIA
Telephone	1300 377 696
Email	sales@liberato.com.au
Website	http://www.liberato.com.au

1.4 Emergency telephone numbers

Emergency

2. HAZARDS IDENTIFICATION

2.1 Classification of the substance or mixture

1300 377 696

CLASSIFIED AS HAZARDOUS ACCORDING TO SAFE WORK AUSTRALIA CRITERIA

Physical Hazards

Not classified as a Physical Hazard

Health Hazards

Skin Corrosion/Irritation: Category 1B Serious Eye Damage / Eye Irritation: Category 1

Environmental Hazards

Not classified as an Environmental Hazard

2.2 GHS Label elements

Pictograms



DANGER

Hazard statements

H314	
H318	

Causes severe skin burns and eye damage. Causes serious eye damage.

Prevention statements

P260	Do not breathe dust/fume/gas/mist/vapours/spray.
P264	Wash thoroughly after handling.
P280	Wear protective gloves/protective clothing/eye protection/face protection/hearing protection.



Response statements P301 + P330 + P331 P303 + P361 + P353 P304 + P340 P305 + P351 + P338 P310 P321 P363	IF SWALLOWED: Rinse mouth. Do NOT induce vomiting. IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water/shower. IF INHALED: Remove person to fresh air and keep comfortable for breathing. IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Immediately call a POISON CENTRE or doctor/physician. Specific treatment is advised - see first aid instructions. Wash contaminated clothing before reuse.
Storage statements P405	Store locked up.
Disposal statements P501	Dispose of contents/container in accordance with relevant regulations.
2.3 Other hazards	

No information provided.

3. COMPOSITION/ INFORMATION ON INGREDIENTS

3.1 Substances / Mixtures

Ingredient	CAS Number	EC Number	Content
PHOSPHORIC ACID	7664-38-2	231-633-2	10 to 30%
AMMONIUM HYDROGEN DIFLUORIDE (AMMONIUM BIFLUORIDE)	1341-49-7	215-676-4	<2%
SULPHURIC ACID	7664-93-9	231-639-5	<2%
2-BUTOXYETHANOL	111-76-2	203-905-0	<1%
HYDROCHLORIC ACID	7647-01-0	231-595-7	0.1 to 1%
WATER	7732-18-5	231-791-2	>60%
ADDITIVE(S)	-	-	Remainder

4. FIRST AID MEASURES

4.1 Description of first aid measures

Еуе	Keep patient calm. Immediately hold eyelids apart and irrigate entire eyeball with gentle flow of water for 15 to 20 minutes. Urgently seek eye specialist attention while continuing irrigation. Calcium gluconate gel may be applied to eyes if medical attention is delayed, or use a dedicated first aid device such as HEXAFLUORINE ® as per supplier's instructions.
Inhalation	Quickly remove from exposure. Remove contaminated clothing, check there is no obstruction to the airway if breathing is weak or has ceased and give artificial respiration, preferably using an oxygen resuscitator. Administer 2.5% calcium gluconate via nebuliser if available. In all cases summon ambulance and transport to hospital for further observation.
Skin	Flush affected area with copious quantities of water. Use an emergency shower for large areas. Remove affected clothing as quickly as possible. Decontaminate with saline or water. Apply calcium gluconate 2.5% gel to contaminated skin, repeating every 15 minutes until pain ceases. If calcium gluconate gel is not available, an extemporaneous gel can be prepared by adding 10 mL of calcium gluconate injection 10% to 30 mL of sterile surgical lubricant. Alternatively, use a dedicated first aid device such as HEXAFLUORINE ® Autonomous Portable Shower as per supplier's instructions. Urgently transport to hospital and recommend admission.
Ingestion	For advice, contact a Poisons Information Centre on 13 11 26 (Australia Wide) or a doctor (at once). If swallowed, do not induce vomiting.
First aid facilities	Emergency shower and eye wash basin. Calcium gluconate gel or injection should also be available and/or dedicated first aid devices such as the HEXAFLUORINE ® range. Rescue personnel should use self contained breathing apparatus and a full chemical suit or full cover overalls. Air-VivaTM or Oxy-VivaTM. Water or sterile saline solution for irrigation.

4.2 Most important symptoms and effects, both acute and delayed

Exposure to high concentrations may result in immediate, severe, burning pain and whitish discoloration proceeding to blister formation. Exposure to lower concentrations may result in pain, redness, swelling and blistering - symptoms may be delayed. Eye exposure may result in severe burns with corneal destruction or opacification. If left untreated, blindness may result. Acute symptoms of inhalation may include coughing, choking, chest tightness, chills, fever and cyanosis. Ingestion may result in severe burns to the mouth, esophagus and stomach. Systemic fluoride toxicity may result from ingestion, inhalation, or extensive dermal burns; hypocalcaemia, hypomagnesemia, hyperkalaemia (potassium), pulmonary oedema, metabolic acidosis, ventricular arrhythmias and death may occur.

4.3 Immediate medical attention and special treatment needed

Treat symptomatically.

5. FIRE FIGHTING MEASURES

5.1 Extinguishing media

Water fog. Prevent contamination of drains and waterways.

5.2 Special hazards arising from the substance or mixture

Non flammable. May evolve toxic gases (fluorides) when heated to decomposition. May evolve flammable hydrogen gas in contact with some metals.

5.3 Advice for firefighters

Evacuate area and contact emergency services. Toxic gases may be evolved in a fire situation. Remain upwind and notify those downwind of hazard. Wear full protective equipment including Self Contained Breathing Apparatus (SCBA) when combating fire. Use waterfog to cool intact containers and nearby storage areas.

5.4 Hazchem code

- 2X
- 2 Fine Water Spray.
- X Wear liquid-tight chemical protective clothing and breathing apparatus. Contain spill and run-off.

6. ACCIDENTAL RELEASE MEASURES

6.1 Personal precautions, protective equipment and emergency procedures

Wear Personal Protective Equipment (PPE) as detailed in section 8 of the SDS. Clear area of all unprotected personnel. Contact emergency services where appropriate.

6.2 Environmental precautions

Prevent product from entering drains and waterways.

6.3 Methods of cleaning up

Contain spillage, then cover / absorb spill with sodium carbonate or similar, collect and place in suitable containers for treatment and/or disposal.

6.4 Reference to other sections

See Sections 8 and 13 for exposure controls and disposal.

7. HANDLING AND STORAGE

7.1 Precautions for safe handling

Before use carefully read the product label. Use of safe work practices are recommended to avoid eye or skin contact and inhalation. Observe good personal hygiene, including washing hands before eating. Prohibit eating, drinking and smoking in contaminated areas.

7.2 Conditions for safe storage, including any incompatibilities

Store in a secured, cool, dry, well ventilated area, removed from incompatible substances, heat or ignition sources and foodstuffs. Ensure containers are adequately labelled, protected from physical damage and sealed when not in use. Check regularly for leaks or spills. Large storage areas should be bunded.

7.3 Specific end uses

No information provided.

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

8.1 Control parameters

Exposure standards

Ingredient	Reference	TWA		STEL	
Ingredient	Kelefence	ppm	mg/m³	ppm	mg/m³
2-Butoxyethanol (EGBE)	SWA [AUS]	20	96.9	50	242
2-Butoxyethanol (EGBE)	SWA [Proposed]	10	49	50	242
Fluorides, as F	SWA [AUS]		2.5		
Hydrogen chloride	SWA [Proposed]	2 (Peak)	2.98 (Peak)		
Hydrogen chloride (Hydrochloric acid)	SWA [AUS]	5 (Peak)	7.5 (Peak)		
Phosphoric acid	SWA [AUS]		1		3
Sulphuric acid	SWA [AUS]		1		3
Sulphuric acid	SWA [Proposed]		0.1		

Biological limits

Ingredient	Reference	Determinant	Sampling Time	BEI
2-BUTOXYETHANOL	ACGIH BEI	Butoxyacetic acid (BAA) in urine (with hydrolysis)	End of shift	200 mg/g creatinine
AMMONIUM HYDROGEN DIFLUORIDE (AMMONIUM BIFLUORIDE)	ACGIH BEI	Fluoride in urine	Prior to shift	2 mg/L
	ACGIH BEI	Fluoride in urine	End of shift	3 mg/L

8.2 Exposure controls

Engineering controls

Avoid inhalation. Use in well ventilated areas. Where an inhalation risk exists, mechanical extraction ventilation is required.

PPE

Eye / Face	Wear a faceshield and splash-proof goggles.
Hands	Wear butyl or Viton® gloves.
Body	Wear rubber boots and impervious coveralls.
Respiratory	Where an inhalation risk exists, wear a Full-face Type B (acid gas) respirator. With prolonged use, wear an Air-line / Full Facepiece Supplied-Air Respirator (SAR).



9. PHYSICAL AND CHEMICAL PROPERTIES

9.1 Information on basic physical and chemical properties

VISCOUS GREEN LIQUID
SLIGHT ODOUR
NON FLAMMABLE
NOT RELEVANT
NOT AVAILABLE
NOT AVAILABLE
NOT AVAILABLE
< 1.0
NOT AVAILABLE
NOT AVAILABLE
SOLUBLE
NOT AVAILABLE
NOT RELEVANT
NOT RELEVANT
NOT AVAILABLE
NOT AVAILABLE

9.1 Information on basic physical and chemical properties

Decomposition temperature	NOT AVAILABLE
Viscosity	NOT AVAILABLE
Explosive properties	NOT AVAILABLE
Oxidising properties	NOT AVAILABLE
Odour threshold	NOT AVAILABLE
9.2 Other information	
% Volatiles	70 % to 90 %
Specific gravity	1.12

10. STABILITY AND REACTIVITY

10.1 Reactivity

Carefully review all information provided in sections 10.2 to 10.6.

10.2 Chemical stability

Stable under recommended conditions of storage.

10.3 Possibility of hazardous reactions

Polymerization will not occur.

10.4 Conditions to avoid

Avoid heat, sparks, open flames and other ignition sources.

10.5 Incompatible materials

Incompatible with oxidising agents (e.g. hypochlorites), acids (e.g. nitric acid), alkalis (e.g. sodium hydroxide) and metals.

10.6 Hazardous decomposition products

May evolve toxic gases (fluorides) when heated to decomposition.

11. TOXICOLOGICAL INFORMATION

11.1 Information on toxicological effects

Acute toxicity

Ingestion may result in severe burns of the mouth and throat, as well as a danger of perforation of the oesophagus and the stomach.

Information available for the ingredients:

Ingredient		Oral LD50	Dermal LD50	Inhalation LC50
PHOSPHORIC ACID		1530 mg/kg (rat)	2740 mg/kg (rabbit)	3846 mg/m³ (rat)
AMMONIUM HYDROGEN DIFLUORIDE (AMMONIUM BIFLUORIDE)		130 mg/kg (rat)		
SULPHURIC ACID		2140 mg/kg (rat)		18 mg/m³ (guinea pig); 510 mg/m3/2hrs (rat)
2-BUTOXYETHANOL		~1200 mg/kg (rat) (ECHA)	220 mg/kg (rabbit)	450 mg/L/4hrs (rat)
HYDROCHLORIC ACID		2210 mg/kg (rat)		1108 ppm/1hr (human - respiratory irritation)
Skin	Contact may result in bu	Contact may result in burning sensation and deep burns with tissue damage.		
Eye Contact may result in irritation, lacrimation, pain, redness, corneal burns and possible serious e		sible serious eye damage.		
Sensitisation Not classified as causing skin or respiratory sensitisation.				

Mutagonioity	Not clossified as a mutagon
Mutagenicity	Not classified as a mutagen.

Not classified as a carcinogen. Occupational exposure to strong inorganic acid mists containing sulphuric Carcinogenicity acid is classified as carcinogenic to humans (IARC Group 1).

Reproductive Not classified as a reproductive toxin.

STOT - single Over exposure may result in mucous membrane irritation of the respiratory tract, coughing, bronchitis, ulceration, bloody nose, lung tissue damage, chemical pneumonitis, and pulmonary oedema. exposure

STOT - repeated Repeated exposure may result in discolouration of teeth; as well as lung, kidney, liver, ligament and bone (osteosclerosis, skeletal fluorosis) damage. exposure

Not classified as causing aspiration. Aspiration



12. ECOLOGICAL INFORMATION

12.1 Toxicity

An environmental hazard cannot be excluded in the event of unprofessional handling or disposal. Phosphoric acid is hazardous to aquatic life at high concentrations.

12.2 Persistence and degradability

While acidity may be reduced by natural water minerals, the phosphate may persist indefinitely.

12.3 Bioaccumulative potential

Not expected to bioaccumulate.

12.4 Mobility in soil

Will permeate downwards and may dissolve some soil matter. Some acid will be neutralised, however significant amounts will remain and may enter the groundwater.

12.5 Other adverse effects

Avoid release to the environment.

13. DISPOSAL CONSIDERATIONS

13.1 Waste treatment methods

Waste disposal For small amounts (as determined by risk assessment or similar): Wearing the protective equipment detailed above, neutralise to pH 6-8 by SLOW addition to a saturated sodium bicarbonate solution or similar basic solution. Dilute with excess water and flush to drain. Waste disposal should only be undertaken in a well ventilated area. For larger amounts: Dispose in accordance with local regulations.

Legislation Dispose of in accordance with relevant local legislation.

14. TRANSPORT INFORMATION

CLASSIFIED AS A DANGEROUS GOOD BY THE CRITERIA OF THE ADG CODE



	LAND TRANSPORT (ADG)	SEA TRANSPORT (IMDG / IMO)	AIR TRANSPORT (IATA / ICAO)
14.1 UN Number	1760	1760	1760
14.2 Proper Shipping Name	CORROSIVE LIQUID, N.O.S. (contains phosphoric acid, sulphuric acid)	CORROSIVE LIQUID, N.O.S. (contains phosphoric acid, sulphuric acid)	CORROSIVE LIQUID, N.O.S. (contains phosphoric acid, sulphuric acid)
14.3 Transport hazard class	8	8	8
14.4 Packing Group	II	II	II

14.5 Environmental hazards

Not a Marine Pollutant.

14.6 Special precautions for user

Hazchem code	2X
GTEPG	8A1
EmS	F-A, S-B

15. REGULATORY INFORMATION

15.1 Safety, health and environmental regulations/legislation specific for the substance or mixture

Poison schedule Classified as a Schedule 6 (S6) Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP).

- **Classifications** Safe Work Australia criteria is based on the Globally Harmonised System (GHS) of Classification and Labelling of Chemicals (GHS Revision 7).
- Inventory listings AUSTRALIA: AIIC (Australian Inventory of Industrial Chemicals) All components are listed on AIIC, or are exempt.

16. OTHER INFORMATION

Additional information HYDROFLUORIC ACID: Severe burns and tissue damage have been reported after direct contact with small quantities of low concentration (< 20 %) hydrofluoric acid. An immediate burning sensation and pain is not always apparent but is a delayed effect which may proceed to corrosive tissue damage and toxic systemic effects through absorption. Hydrofluoric acid has the potential to cause permanent tissue damage and to be fatal if contaminated areas are not treated immediately.

PREHOSPITAL CARE:

Treatment for HF acid burns includes basic life support and appropriate decontamination, followed by neutralisation of the acid by use of calcium gluconate. If exposure occurs at an industrial site, obtain and transport any available treatment literature.

1. Assess and manage acute life threatening conditions in the usual manner. Emergency Medical Services (EMS) personnel should use gloves, masks, and gowns, if necessary.

2. Remove contaminated clothing. Flush with copious amounts of water.

3. Ice packs on the affected area may alleviate symptoms by retarding diffusion of the fluoride ion.

4. If calcium gluconate gel is available, apply liberally to the affected area.

5. For digital burns, if calcium gluconate gel is not available, the fingers may be soaked in magnesium hydroxide-containing antacid preparations (eg, Mylanta) en route to a medical facility.

- 6. Treat inhalation exposures with oxygen and 2.5% calcium gluconate nebuliser.
- 7. Transport the patient to the nearest appropriate medical facility.

(Reference: eMedicine Journal, May 7 2001, Volume 2, Number 5).

PERSONAL PROTECTIVE EQUIPMENT GUIDELINES:

The recommendation for protective equipment contained within this report is provided as a guide only. Factors such as form of product, method of application, working environment, quantity used, product concentration and the availability of engineering controls should be considered before final selection of personal protective equipment is made.

HEALTH EFFECTS FROM EXPOSURE:

It should be noted that the effects from exposure to this product will depend on several factors including: form of product; frequency and duration of use; quantity used; effectiveness of control measures; protective equipment used and method of application. Given that it is impractical to prepare a report which would encompass all possible scenarios, it is anticipated that users will assess the risks and apply control methods where appropriate.



Abbreviations	ACGIH	American Conference of Governmental Industrial Hygienists		
	CAS #	Chemical Abstract Service number - used to uniquely identify chemical compounds		
	CNS	Central Nervous System		
	EC No.	EC No - European Community Number		
	EMS	Emergency Schedules (Emergency Procedures for Ships Carrying Dangerous Goods)		
	GHS	Globally Harmonized System		
	GTEPG	Group Text Emergency Procedure Guide		
	IARC	International Agency for Research on Cancer		
	LC50	Lethal Concentration, 50% / Median Lethal Concentration		
	LD50	Lethal Dose, 50% / Median Lethal Dose		
	mg/m³	Milligrams per Cubic Metre		
	OEL	Occupational Exposure Limit		
	рН	relates to hydrogen ion concentration using a scale of 0 (high acidic) to 14 (highly alkaline).		
	ppm	Parts Per Million		
	STEL	Short-Term Exposure Limit		
	STOT-RE	Specific target organ toxicity (repeated exposure)		
	STOT-SE	Specific target organ toxicity (single exposure)		
	SUSMP	Standard for the Uniform Scheduling of Medicines and Poisons		
	SWA	Safe Work Australia		
	TLV	Threshold Limit Value		
	TWA	Time Weighted Average		
Report status		It has been compiled by RMT on behalf of the manufacturer, importer or supplier of the erves as their Safety Data Sheet ('SDS').		
	manufacturer, the current sta at the time of	on information concerning the product which has been provided to RMT by the importer or supplier or obtained from third party sources and is believed to represent ate of knowledge as to the appropriate safety and handling precautions for the product f issue. Further clarification regarding any aspect of the product should be obtained he manufacturer, importer or supplier.		
	While RMT has taken all due care to include accurate and up-to-date information in this SDS, it does not provide any warranty as to accuracy or completeness. As far as lawfully possible, RMT accepts no liability for any loss, injury or damage (including consequential loss) which may be suffered or incurred by any person as a consequence of their reliance on the information contained in this SDS.			
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